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For the levels to differ the pressure  $P_1$  must be greater than  $P_2$ , hence.  $P_1 = P_2 + h\rho g$ . If  $P_1$  is the lung pressure,  $P_0$  is the atmospheric pressure, then if the difference is 'h' then lung pressure can be calculated as follows.  $P_1 = P_0 + h\rho g$ . Example. A man blows into one end of a U-tube containing water until the levels differ by 40.0 cm. if the atmospheric pressure is  $1.01 \times 10^5$  ...

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